

Wafer-Scale Production of Transition Metal Dichalcogenides and Alloy Monolayers by Nanocrystal Conversion for Large-Scale Ultrathin Flexible Electronics

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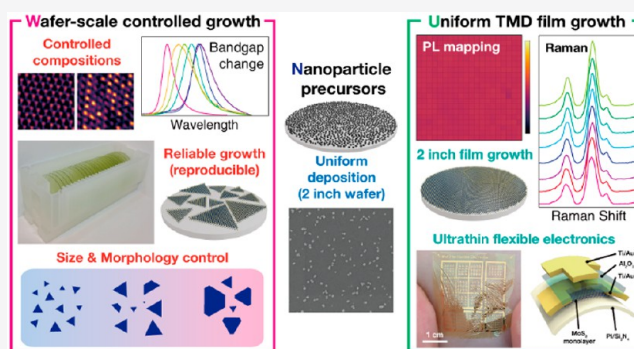
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ABSTRACT: Two-dimensional (2D) transition metal dichalcogenide (TMD) layers are unit-cell thick materials with tunable physical properties according to their size, morphology, and chemical composition. Their transition of lab-scale research to industrial-scale applications requires process development for the wafer-scale growth and scalable device fabrication. Herein, we report on a new type of atmospheric pressure chemical vapor deposition (APCVD) process that utilizes colloidal nanoparticles as process-scalable precursors for the wafer-scale production of TMD monolayers. Facile uniform distribution of nanoparticle precursors on the entire substrate leads to the wafer-scale uniform synthesis of TMD monolayers with the controlled size and morphology. Composition-controlled TMD alloy monolayers with tunable bandgaps can be produced by simply mixing dual nanoparticle precursor solutions in the desired ratio. We also demonstrate the fabrication of ultrathin field-effect transistors and flexible electronics with uniformly controlled performance by using TMD monolayers.

KEYWORDS: transition metal dichalcogenides, nanoparticle precursor, wafer-scale growth, transition metal dichalcogenide alloys, ultrathin flexible electronics



Transition metal dichalcogenide (TMD) materials with controlled compositions are attractive materials for electronics and catalysis.^{1–3} This is because atom-thick TMD monolayers present unique properties in terms of exciton dynamics, the valley Hall effect, and the indirect-to-direct bandgap transition,^{4–7} different from their bulk counterparts. In addition, varying the size and morphology of TMDs provides opportunities to tune the reactivity, which further enhances their potential in practical applications.^{8–10} However, challenges in wafer-scale production of uniform TMD monolayers with respect to the size, morphology, and composition restrict the transformation from lab-scale research to industry-level applications in electrocatalysis, electronics, and optoelectronics.

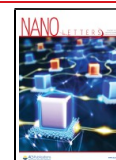
In recent years, diverse TMD preparation methods, such as mechanical exfoliation, liquid exfoliation, and vapor–solid growth,^{11–14} have been developed. Among them, chemical vapor deposition (CVD) is the most compatible method with wafer-scale processing of 2D materials. Although metal–organic CVD (MOCVD) is a typical method for the TMD growth,^{15,16} it requires several hours of complicated processes

and the use of metal–organic precursors, which are toxic and expensive. Atmospheric pressure CVD (APCVD) is an attractive alternative, because it facilitates scalable TMD growth in simple and cost-effective manner.^{17,18} Nevertheless, the control of size, morphology, and composition of the TMD monolayers is hampered by difficulties in the stable and ratio-controlled supply of constituent monomers by sublimation and vaporization of multiple powder precursors. Thus, the key challenge for the desired wafer-scale growth of TMD monolayers is how to achieve uniform nucleation and growth of the layers in a short time throughout the substrate by using precursors with sufficient reactivity, nontoxicity, and wafer-process compatibility.

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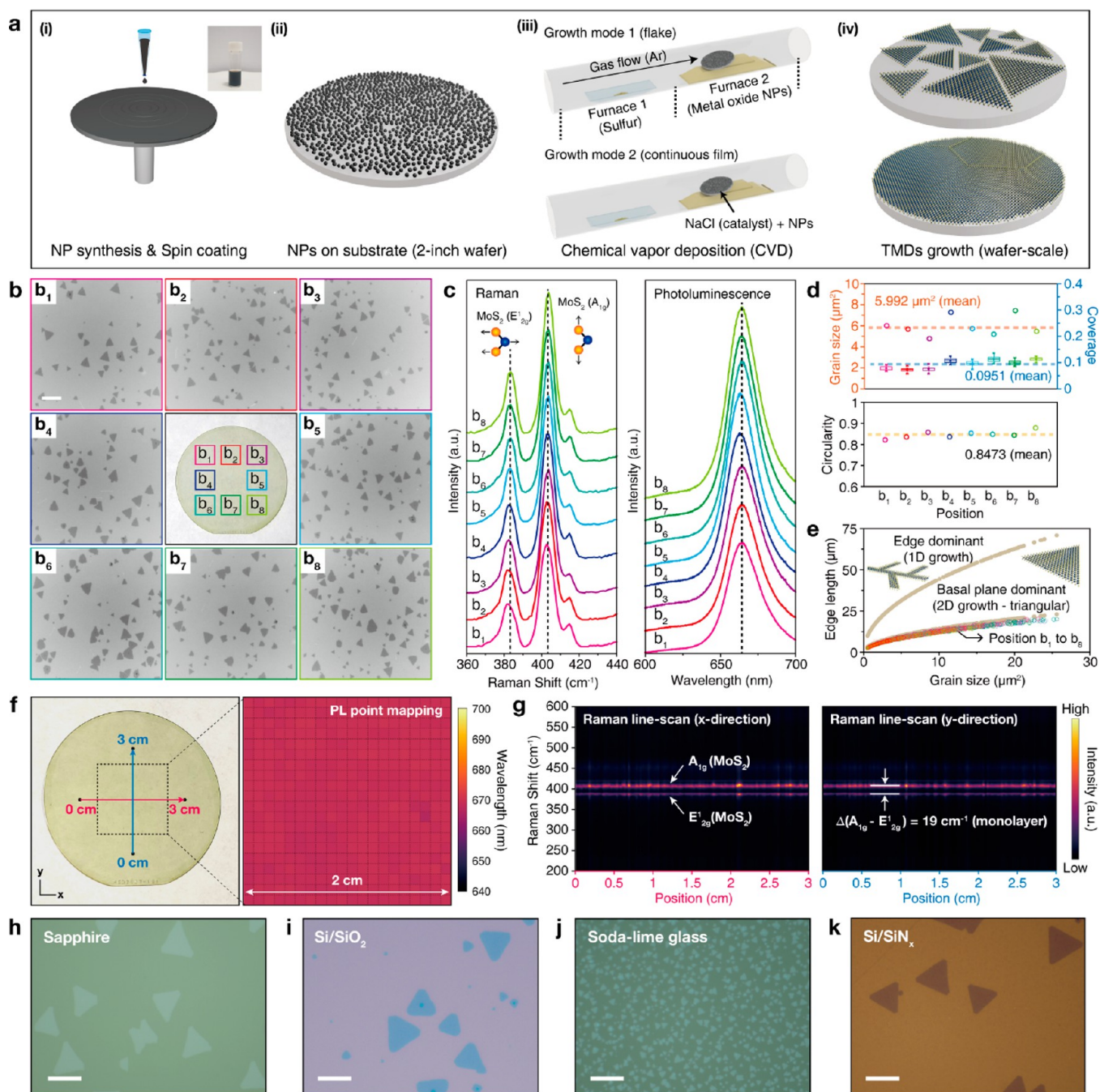


Figure 1. Uniform synthesis of MoS₂ monolayer flakes and continuous film. (a) Schematic of the entire synthetic process. (b) Photograph of a 2 in. wafer after the synthetic process and OM images of MoS₂ layers from positions b₁ to b₈ of the 2 in. wafer. Scale bar, 10 μm. (c) Raman and PL spectra of MoS₂ monolayers from positions b₁ to b₈. Dashed lines indicate the characteristic peaks of MoS₂ monolayer. (d) Averaged grain size, coverage, and circularity of thousands of MoS₂ layers from positions b₁ to b₈. (e) Plot of edge length and grain size of individual MoS₂ flakes from positions b₁ to b₈. Schematics and beige dots indicate ideal branched and triangular shapes, respectively. (f) Photograph of a 2 in. wafer covered by continuous MoS₂ film and PL spectral point mapping data of the film. (g) Raman spectra (line-scan along *x* (red arrow) and *y* (blue arrow) directions shown in part f) of the continuous film. (h–k) OM images of MoS₂ monolayer flakes grown on sapphire, Si/SiO₂, soda-lime glass, and Si/SiN_x wafers. Scale bars, 10 μm.

Herein, we develop a versatile APCVD process for wafer-scale controlled growth of TMD monolayers based on the conversion chemistry of nanoparticle precursors. Attributed to the colloidal stability and high reactivity of nanoparticles, uniform distribution of a minimum amount of precursors and homogeneous and fast growth of TMD monolayers by solid–vapor–solid transition on inch-scale substrates are accomplished. This method is readily extended to the wafer-scale

TMD alloy growth with varied compositions, such as Mo_{1–x}W_xS₂, with a tunable bandgap energy, simply by using mixed nanoparticle precursors. Moreover, the uniform MoS₂ film grown by our method enables the large-scale fabrication of field-effect transistors and their application to ultrathin flexible electronics with uniform performance over the entire substrate.

The overall process for the wafer-scale TMD growth is described in Figure 1a. Monodisperse transition metal oxide

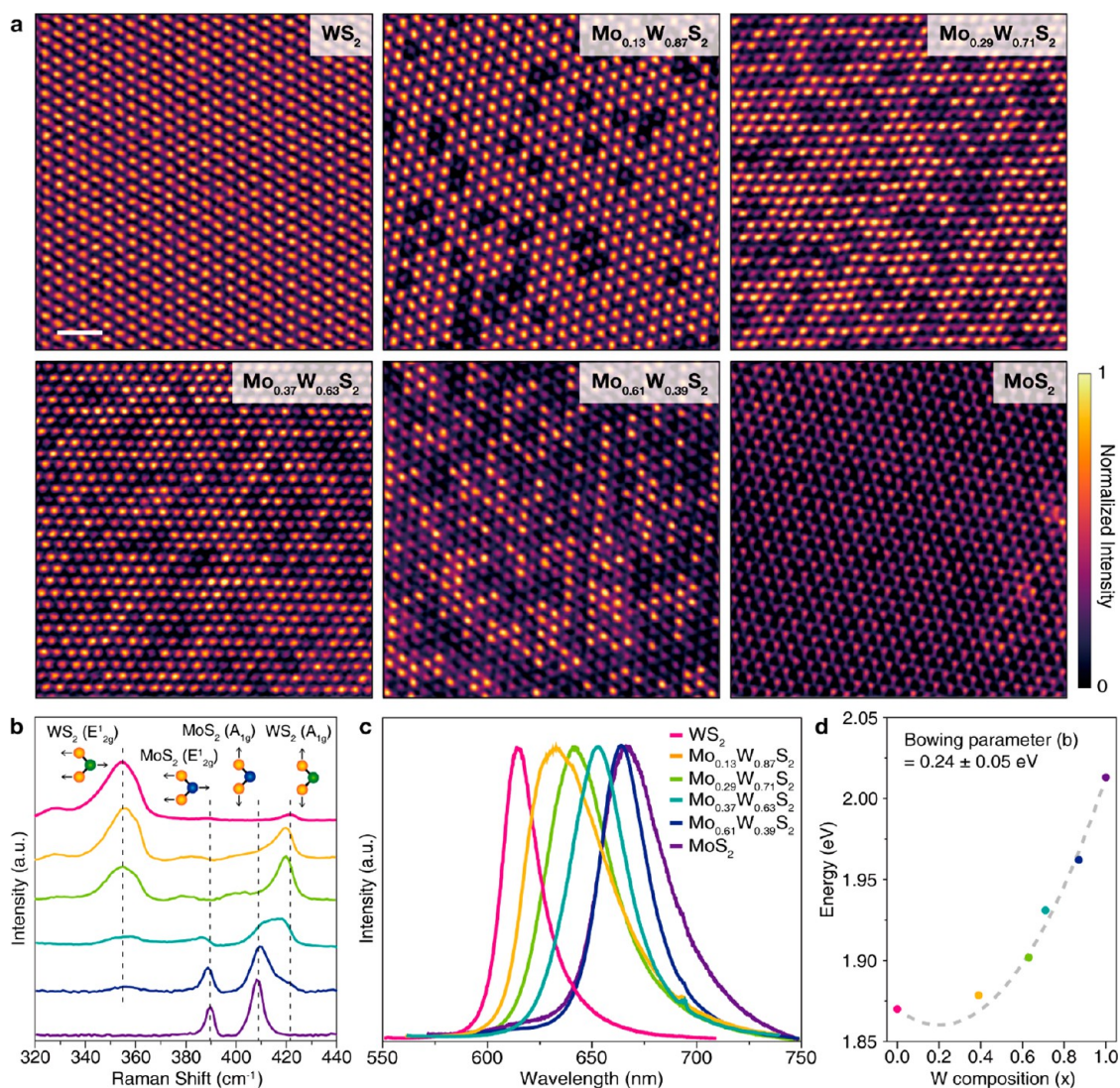


Figure 2. Synthesis and characterizations of $\text{Mo}_{1-x}\text{W}_x\text{S}_2$ alloy monolayers. (a) FFT-filtered HAADF-STEM images of $\text{Mo}_{1-x}\text{W}_x\text{S}_2$ monolayers. Scale bar, 1 nm. Intensities are normalized by MATLAB code, and images are color-mapped by normalized intensity (right, color bar). (b) Raman spectra of $\text{Mo}_{1-x}\text{W}_x\text{S}_2$ monolayers. Schematics (inset) show the vibrational modes of MoS_2 and WS_2 monolayers. Dashed lines indicate the characteristic Raman peaks of MoS_2 and WS_2 monolayers. (c) PL spectra of $\text{Mo}_{1-x}\text{W}_x\text{S}_2$ monolayers with the controlled compositions. (d) Plot of the measured optical bandgap energies of $\text{Mo}_{1-x}\text{W}_x\text{S}_2$ monolayers (x , 0–1). Calculated bowing parameter is (0.24 ± 0.05) eV.

nanoparticles, such as MoO_2 and $\text{W}_{18}\text{O}_{49}$, are prepared in a large scale by colloidal synthesis (Figure S1) and stored in a solution as precursors for TMD synthesis.^{19,20} The average size of the MoO_2 nanoparticles is 56.36 nm (Figure S2). The precursors are deposited on a 2 in. wafer by spin coating, which is facilitated by the colloidal stability of the nanoparticles. To obtain the desired coverage of the nanoparticles, the concentration of the solution and the rotational speed of the spin coating are controlled (Figure S3). A uniform distribution of nanoparticles over the substrate is easily achieved without any thermal and high-vacuum systems typically used for pretreatment of the precursors, confirmed by scanning electron microscopy (SEM) (Figure S4). The substrate with the loaded nanoparticles and sulfur powder are inserted into two different furnaces in CVD for independent heating (Figure S5). The CVD process on a substrate with a tilted angle toward the direction of carrier gas (growth mode 1 in Figure 1a) provides a uniform supply of the sulfur precursor and yields TMD monolayers with controlled density.

As a representative example of growth mode 1, the wafer-scale growth of MoS_2 is demonstrated, and confirmed by the consistency in the optical microscopy (OM), Raman scattering, and photoluminescence (PL) spectra that are obtained from different positions of the substrate, marked as b_1 – b_8 (Figure 1b). The Raman spectra of the layers from multiple positions of the substrate, excited by a 532 nm laser, are uniform and exhibit that the difference of two characteristic peaks (E'_{2g} and A_{1g}) of the MoS_2 monolayer is approximately 19 cm^{-1} (Figure 1c), which is consistent with the reported value.²¹ Strong and uniform peaks in the PL spectra also indicate that MoS_2 layers are monolayered and highly crystalline.^{22,23} Furthermore, the projected area and perimeter of thousands of MoS_2 monolayers throughout the wafer are extracted by image processing of OM images using home-built MATLAB codes. The averaged grain size and coverage of the layers from different positions are uniform and close to $5.992 \mu\text{m}^2$ and 0.0951, which are the mean grain size and coverage over the inch-scale wafer, respectively (Figure 1d). We also

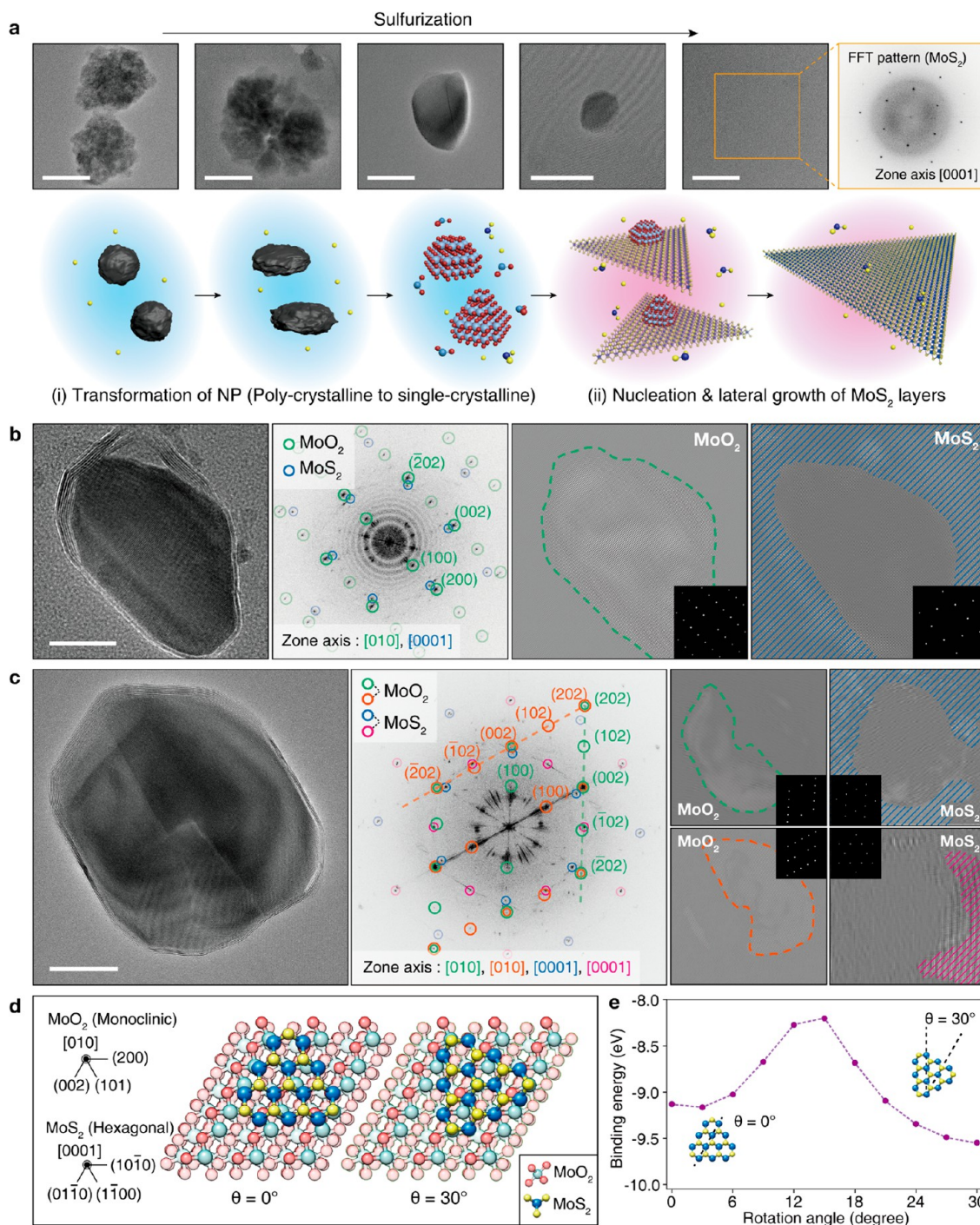


Figure 3. MoO₂ transformation and MoS₂ growth processes. (a) HRTEM images and schematics of MoO₂ nanoparticles and underlying grown MoS₂ with the progress of sulfurization. Scale bars, 50 nm. (b, c) HRTEM images, corresponding FFT patterns, and inverse FFT patterns of MoO₂ crystals and MoS₂ domains. Scale bars, 20 nm. MoO₂ and MoS₂ are marked by (green and orange) and (blue and magenta) circles, respectively. Insets are masked FFT patterns for each crystal. (d) Atomic models for MoO₂ and MoS₂ crystals arranged with rotation angles of (0 and 30)^o. MoO₂ and MoS₂ domains are aligned to zone axis (Z.A.) [010] and [0001], respectively. (e) DFT calculation of binding energy between MoS₂ and the underlying MoO₂ crystals with different rotation angles.

investigate the morphological uniformity of the layers by measuring their circularity. The average circularity of the layers from different positions is close to 0.8473 (mean value). Furthermore, the plots between the grain size and the edge length measured in positions b₁–b₈ consistently follow the trend expected for the ideal triangular monolayer (Figure 1e). These results demonstrate the uniform growth of triangular monolayers throughout the substrate. To examine the epitaxial

behavior between as-grown MoS₂ and sapphire substrate, orientations of MoS₂ layers from the positions b₁ to b₈ are extracted by using MATLAB codes (Figure S6). Histograms in Figure S6 present that randomly oriented MoS₂ monolayers are grown in our process, although the grains aligned 30^o and 90^o with the [1100] direction of the sapphire substrate are dominant. This is because the sulfur precursor is moderately supplied in our growth process, and the alignment can be

further tuned by controlling the supply rate of the sulfur precursor.²⁴ Nanoparticle precursors mixed with NaCl (growth mode 2 in Figure 1a) promote the growth of the extended TMD film, due to the catalytic effect of sodium.^{25–27} Due to the large surface-to-volume ratio and resulting high reactivity of the nanoparticles (Table S1), only 0.608 mg of MoO₂ and 0.152 μg of NaCl for MoS₂ are used for the continuous film growth on the 2 in. wafer. The uniformity of PL spectral point mapping (400 points; point interval, 1 mm) and Raman line-scan data (120 points along the *x* and *y* directions; point interval, 0.25 mm) indicates that a continuous MoS₂ monolayer film is uniformly grown over the entire substrate (Figure 1f,g). Coverage of the MoS₂ continuous film is measured by SEM, demonstrating the near full coverage throughout the entire wafer substrate (Figure S7). Merged grains in the film are coherently connected without overlapping, confirmed by dark-field TEM imaging (Figure S8). Grain size of the film is generally 10 μm with a small population of small domains (~1 μm). Grain boundary structures are also investigated by Cs-corrected STEM analysis (Figure S9). We confirm that the merged grains are well stitched at the boundaries, regardless of their crystallographic orientations and the presence of a small portion of the incoherent boundary.^{28–30} The MoS₂ growth from MoO₂ nanoparticles is also demonstrated for other types of substrates, including Si/SiO₂, soda-lime glass, and Si/SiN_x, proving the versatility of nanoparticle precursors (Figure 1h–k).

Compositional control of the TMD monolayer is a versatile method of manipulating the electronic structure and tuning the bandgap energy.^{31–35} However, in the conventional CVD process, it is difficult to precisely control the stoichiometry of the TMD monolayer because the multiple powder precursors are loaded in different chambers whose thermal conditions are independently controlled, and the growth relies on their complex sublimation and transport processes. We demonstrate that our method can overcome this problem. Because the metal precursors are directly supplied on the substrate where the TMD monolayer is to grow, loading two different nanoparticle precursors facilitates improved compositional control of the TMD alloys. We synthesize Mo_{1–*x*}W_{*x*}S₂ (*x*, 0–1) monolayers by using a dual-precursor of molybdenum oxide and tungsten oxide nanoparticle solutions that are easily mixed and uniformly deposited on the substrate, attributing to the colloidal stability of the mixed nanoparticle precursors. Simple mixing of the dual-precursor with different concentrations (Table S2) allows the Mo_{1–*x*}W_{*x*}S₂ alloy growth with controlled compositions. High-angle annular dark-field-scanning transmission electron microscopy (HAADF-STEM) is employed to visualize the arrangement of Mo and W atoms in Mo_{1–*x*}W_{*x*}S₂ monolayers with the compositional variation, where the intensity is dependent on the atomic number³⁶ (Figure 2a). The ratio between W and Mo is estimated by counting the number of bright (W) and dark (Mo) atoms from multiple atomic-resolution HAADF-STEM images.

Raman spectra of the WS₂ and MoS₂ monolayers show the characteristic peaks of E_{2g}¹ and A_{1g} for both materials^{37–39} (Figure 2b). Mo_{1–*x*}W_{*x*}S₂ alloy monolayers show the characteristic peaks of both MoS₂ and WS₂ with changing relative intensities and peak positions between them, which correspond to the metallic composition. Most notably, PL measurements confirm that the bandgap energy is successfully tuned by the synthesis of Mo_{1–*x*}W_{*x*}S₂ alloys with the controlled composi-

tions (Figure 2c). The bandgap energies of the Mo_{1–*x*}W_{*x*}S₂ alloy monolayers match well with the bowing effect (Figure 2d), following the given eq 1:

$$E_{\text{PL,Mo}_{1-x}\text{W}_x\text{S}_2} = (1-x)E_{\text{PL,MoS}_2} + xE_{\text{PL,WS}_2} - bx(1-x) \quad (1)$$

where *x* is the tungsten (W) composition; $E_{\text{PL,MoS}_2}$, $E_{\text{PL,WS}_2}$, and $E_{\text{PL,Mo}_{1-x}\text{W}_x\text{S}_2}$ are the measured optical bandgap energies of MoS₂, WS₂, and Mo_{1–*x*}W_{*x*}S₂ monolayers, respectively; and *b* is the calculated bowing parameter of (0.24 ± 0.05) eV, which is the same as the reported value.^{40,41} These results prove that the bandgap energy of the Mo_{1–*x*}W_{*x*}S₂ alloys is successfully tuned by the compositional change of randomly distributed atoms by using a dual-precursor.

We investigate the role of MoO₂ nanoparticle precursors for MoS₂ growth by using high-resolution transmission electron microscopy (HRTEM) of different growth stages. MoS₂ and MoO₂ nanocrystals from the different stages are transferred onto TEM grids by the surface-energy-assisted transfer method,⁴² and imaged by HRTEM (Figure 3a). In the early stage of sulfurization, polycrystalline MoO₂ nanoparticles spread out and increase the size in the lateral direction of the substrate and continuously transform into nanocrystals with single or twinned domains (Figure S10a). MoS₂ monolayer is then nucleated and grows at the bottom of the MoO₂ nanoparticle that acts as a self-seeding material.^{43,44} During the sulfurization, MoO₂ nanoparticles are constantly consumed by adjacently growing MoS₂ monolayers, which is indicated by the diminishing size of the particles. Cross-sectional view of the intermediate MoO₂ particles with underlying MoS₂ monolayer indicates the pancakelike morphology of the intermediate particle (Figure S10b). After exhaustion of the particle, MoS₂ monolayers are synthesized on the substrate, confirmed by fast Fourier transform (FFT).

For further understanding of how MoO₂ nanoparticles determine the growth process of MoS₂ monolayers, HRTEM images of the intermediate stages where both materials coexist are analyzed. We intentionally use an excess supply of sulfur as a means to protect and capture the intermediate MoO₂ particles by covering them with the multilayer MoS₂ shell. It facilitates further HRTEM analysis of the intermediate stages during the growth. The HRTEM image in Figure 3b shows a single-crystalline MoO₂ nanoparticle with the underlying MoS₂ monolayer. Indexing the FFT pattern based on the measured distances of lattice planes and their relative angles indicates the monoclinic MoO₂ nanocrystal aligned into the [010] zone axis (Z.A.). Interestingly, the {10 $\bar{1}$ 0} planes of the MoS₂ strongly correlate with the (002), (200), and ($\bar{2}$ 02) planes of monoclinic MoO₂ (green–blue paired circles). Spatial arrangement of the MoO₂ nanoparticle and the growing MoS₂ monolayer is visualized by inverse FFT obtained from independently masked FFT patterns for the two materials (inset). In addition, FIB (focused ion beam)-HRTEM analysis presents that (020) planes of the intermediate MoO₂ particle (nucleation seed) are aligned with the (0001) plane of the sapphire (Figure S11), which is consistent with Figure 3b. The MoS₂ growth from a MoO₂ nanoparticle composed of twinned domains further confirms that the growth behavior of MoS₂ is determined by the MoO₂ nanoparticle (Figure 3c). Two domains of the MoO₂ nanocrystal are independently correlated with two MoS₂ layers grown adjacently. Both domains of the twinned MoO₂ nanocrystal expose specific planes along the

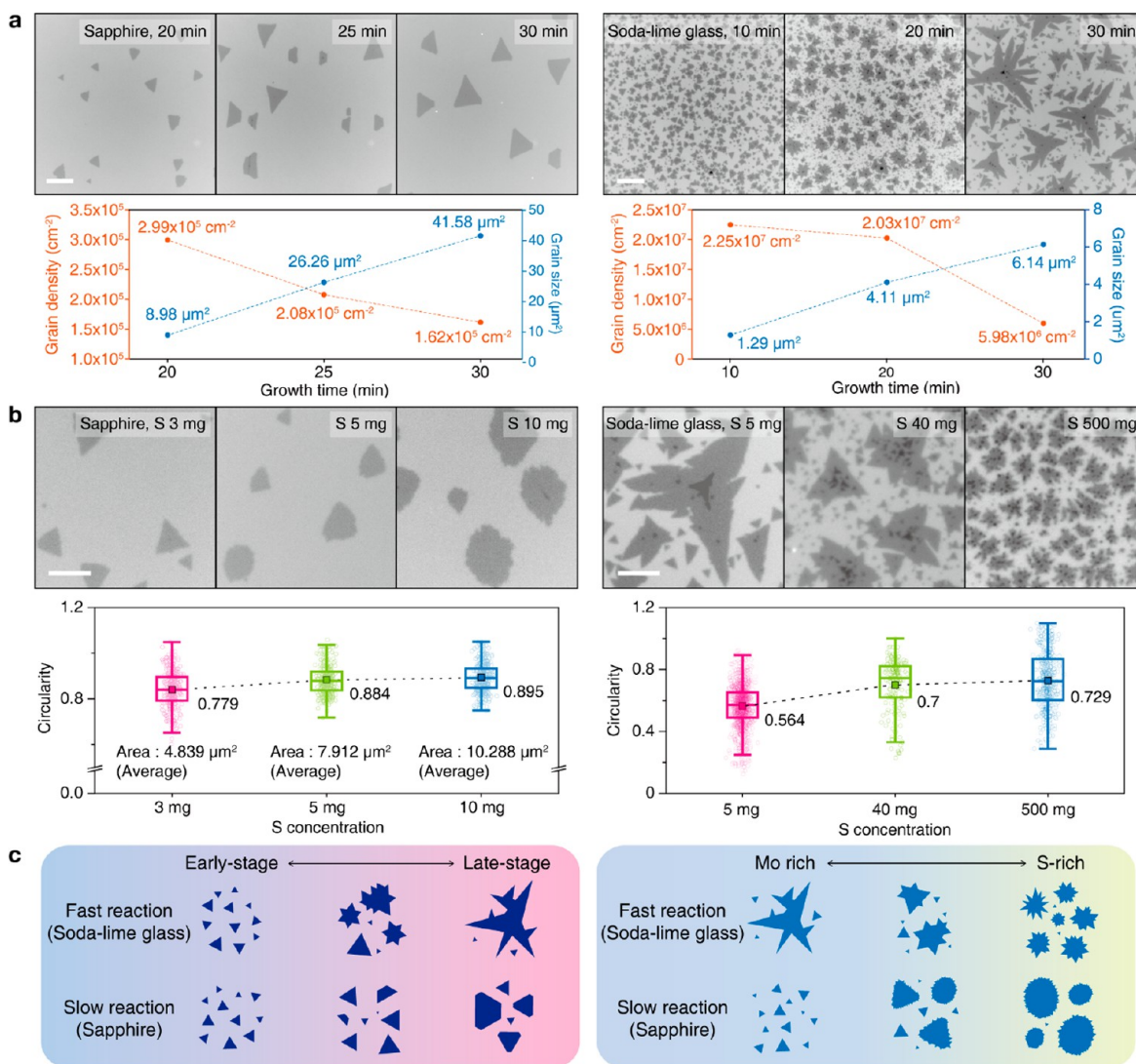


Figure 4. Controlled synthesis of MoS₂ monolayers with various growth parameters. (a) OM images of MoS₂ monolayers grown on sapphire and soda-lime glass wafer with different growth times. Plots show the change of grain density and grain size of the MoS₂ monolayer with respect to growth time. Scale bars, 10 μm. (b) OM images of MoS₂ monolayers grown on sapphire and soda-lime glass wafer with varying S concentrations. Averaged circularities of MoS₂ monolayers are shown on the plot, which indicates the change of morphology of MoS₂ layers. Scale bars, 10 μm. (c) Schematic illustrations of the size and shape evolution of MoS₂ monolayers by changing the growth parameters with different kinetic controls.

Z.A. [010] (Figure S12). The (002), (200), and ($\bar{2}$ 02) planes of one MoO₂ domain with the Z.A. [010] are associated with the orientation of the underlying MoS₂ monolayer in a consistent way, as observed in Figure 3b (green–blue paired circles). The orientation of the MoS₂ monolayer from another twinned MoO₂ domain with the Z.A. [010] rotated by 30° (orange–magenta paired circles). Inverse FFT patterns of the two domains of MoO₂ and the independently grown MoS₂ monolayers represent their arrangements. Furthermore, density functional theory (DFT) calculation reveals energetically preferred orientations in the growth of MoS₂ {1010} on MoO₂ (200) (Figure 3d,e). A local minima of the binding energy are observed at around (0 and 30)° rotation angles, which is consistent with the experimental data.

The functionality of the TMD monolayer in various applications can be improved by controlling the size and morphology.^{45–47} We demonstrate that the size and shape of MoS₂ monolayers can be easily tuned by manipulating the growth parameters, confirmed by quantitative analysis of MoS₂

layers from different parameters that include the growth time, precursor concentration, and substrate type. With prolonged growth time, MoS₂ on sapphire increases the average grain size while maintaining triangular morphologies with marginal truncation of corners, and the proportion of small grains is gradually reduced, as indicated by the OM images and plots in Figure 4a. The MoS₂ growth on soda-lime glass is initiated by densely packed nucleation sites, possibly due to the catalytic characteristic of the substrate.²⁷ The growth results in multiple-branched morphology with the increased size, as it is highly kinetically controlled (Figure 4a). In both cases, ripening of preformed domains occurs in the late stage of sulfurization, since the monomer concentration is locally limited, which allows the consumption of unstable small grains by adjacently growing monolayers. Furthermore, the ratio of molybdenum to sulfur manipulates the edge structure of the layer, resulting in different morphologies. The shape evolution associated with the sulfur concentration is investigated with OM images and circularity. The OM images and measured

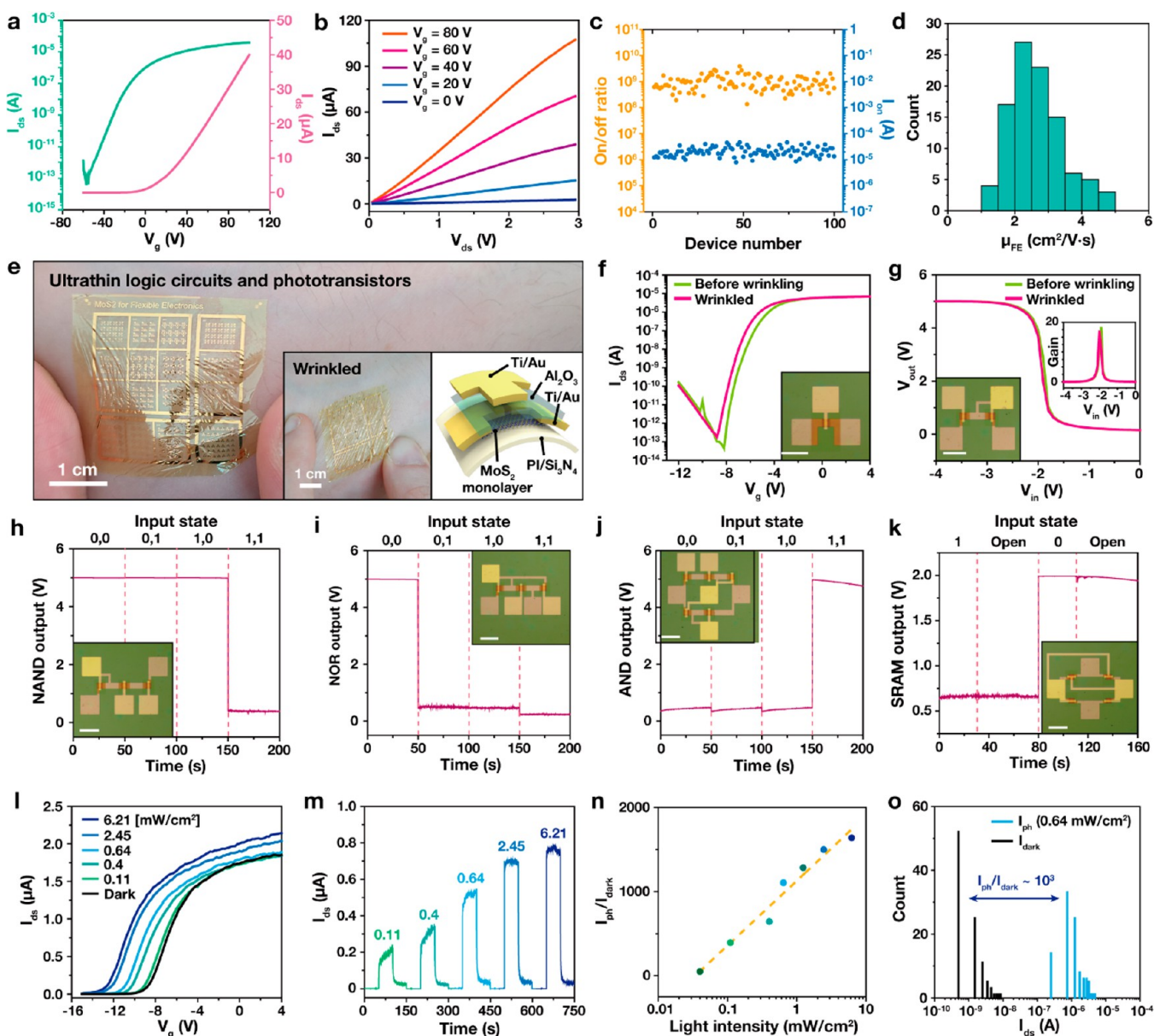


Figure 5. Characterization of MoS₂ transistors, ultrathin logic circuits, and phototransistors. (a) Transfer curves at $V_{ds} = 1$ V. (b) Output characteristics. (c, d) The distribution of on/off ratio, on-current (c, left and right axis), and field-effect mobility (d) measurements from 100 random devices. (e) Photograph of the ultrathin logic circuits and phototransistors attached on human skin. The left inset shows a photograph of the ultrathin device wrinkled on human skin. The right inset shows a schematic of ultrathin MoS₂ FETs. (f, g) Transfer curves of the ultrathin MoS₂ FET (f) and inverter (g) before wrinkling (green) and after wrinkling (pink). $V_{dd} = 5$ V. Inset shows the voltage gain. Scale bars (inset), 200 μ m. (h–k) Output characteristics of NAND (h), NOR (i), and AND gates (j) at $V_{dd} = 5$ V and SRAM (k) at $V_{dd} = 2$ V. Scale bars (Inset), 200 μ m. (l) Transfer curves under the dark and different light intensities (532 nm). (m) Time-resolved photocurrents under different light intensities at $V_{ds} = 1$ V and $V_g = -10$ V. (n) The ratio of photocurrent to dark current under different light intensities. (o) The distribution of dark current and photocurrent under A light intensity of 0.64 mW/cm² measured from a 10 \times 10 phototransistor array.

circularities, shown in Figure 4b, show that in high sulfur concentrations, sharp triangular layers become round-shaped layers with rough edge interfaces. The effect of the ratio of the two precursors is more obvious in the MoS₂ growth on soda-lime glass. The molybdenum-dominant environment induced by the limited sulfur supply promotes the growth of layers with sharp edges, indicated by the low circularity (0.564). In contrast, with an excess of sulfur, MoS₂ monolayers show highly dendritic morphologies with relatively homogeneous size distribution, presumably by suppression of the ripening process.^{48,49} The schematics of Figure 4c summarize the size

and shape evolution process of the MoS₂ monolayer on different substrates.

The uniform wafer-scale synthesis of the MoS₂ film offers great potential for high-performance electronic and optoelectronic devices, including flexible electronics.^{50,51} To demonstrate such potential, the as-grown MoS₂ film is transfer-printed from a sapphire wafer to various desired substrates (e.g., silicon oxide wafer, flexible polyimide substrate), and various devices are fabricated. For the transfer printing, our modified transfer method is applied,⁴² facilitating the intact transfer of the MoS₂ film onto the desired substrate without any noticeable wrinkles (Figure S13).

In order to investigate the uniformity of the MoS₂ film, we fabricate a batch of MoS₂ field-effect transistors (FETs) (3078 transistors shown in Figure S14a). The MoS₂ FET has a bottom-gate structure (Figure S14b). A brief description of the fabrication procedure is as follows. First, the MoS₂ film is transferred onto a 2 in. SiO₂/Si wafer (300 nm thick SiO₂ layer on a Si wafer). Then, the film is patterned by photolithography and dry etching. Afterward, Ti/Au layers (5 nm/25 nm) are deposited by thermal evaporation and patterned by the lift-off process to form source/drain electrodes. Finally, Al₂O₃ is deposited by the atomic layer deposition (ALD) process to form a passivation layer (20 nm). Transfer curves ($I_{ds}-V_g$) and output current characteristics ($I_{ds}-V_{ds}$) of a representative transistor (channel length and width, 5 and 60 μm) are shown in Figure 5a,b. The key device features include the field-effect mobility (μ_{FE}) of 4.13 $\text{cm}^2/\text{V s}$, off-current (I_{off}) of 37 fA, on-current (I_{on}) of 39.9 μA , and an on/off ratio of 10^9 . The device characterization data, such as I_{on} , I_{off} , on/off ratio, μ_{FE} , and threshold voltage (V_{TH}), are collected from 100 randomly selected transistors. The transfer curves of the transistors are shown in Figure S14c,d. Figure 5c shows the on/off ratio (left axis) and I_{on} (right axis) with a mean value of 10^9 and 21.9 μA , respectively. Figure 5d shows the μ_{FE} data, whose average, standard deviation, and maximum are 2.64 $\text{cm}^2/\text{V s}$, 0.8 $\text{cm}^2/\text{V s}$, and 5 $\text{cm}^2/\text{V s}$, respectively. The threshold voltages are mostly in the range of 0–5 V (Figure S14e).

Based on the uniform MoS₂ FETs, we fabricate an ultrathin flexible type of logic circuit (e.g., inverter, NAND gate, NOR gate, AND gate, and SRAM) and phototransistors. Due to their ultrathin flexible nature, the devices have the potential for skin-mounted electronics (Figure 5e). Detailed layouts of the circuits and phototransistor are shown in Figure S15. As illustrated in Figure 5e (right inset), the ultrathin MoS₂ FET has a top-gate structure that consists of polyimide/silicon nitride (PI/Si₃N₄) substrate (420 nm/25 nm), MoS₂ monolayer channel, Ti/Au source/drain electrodes (5 nm/25 nm), Al₂O₃ dielectric layer (25 nm), Ti/Au gate (5 nm/25 nm), and parylene film for the top encapsulation (500 nm). Due to the intrinsic softness of the MoS₂ film and the ultrathin device thickness ($\sim 1 \mu\text{m}$), the devices are tolerable against external mechanical deformations (Figure 5e, left inset). For example, the transfer characteristics show minimal changes under its mechanical deformation (Figure 5f). A transfer curve ($V_{in}-V_{out}$) of the inverter is shown in Figure 5g. The inverter shows a voltage gain of 18.5 at $V_{dd} = 5 \text{ V}$, and stable operation, even under its wrinkled state. Other types of logic gates are also fabricated by using the ultrathin FETs and inverters as building blocks (Figure S16). The NAND gate and NOR gate consist of a transistor and an inverter (Figure 5h,i). The AND gate is the combination of a NAND gate and an inverter (Figure 5j). These logic gates exhibit the output logic state of either “0” (0 V) or “1” (5 V), according to four states of two inputs (e.g., (0,0), (0,1), (1,0), and (1,1)) with $V_{dd} = 5 \text{ V}$. SRAM, which is a flip-flop memory cell, consists of two cross-coupled inverters (Figure 5k). When the input is set to “1” at 0 s or “0” at 80 s, the logic state becomes “0” or “1”, respectively. The logic state can be maintained for 50 s, even after the input is opened at 30 or 110 s, exhibiting stable memory performance. The reliable operation of logic gates corroborates the potential of the MoS₂ film for applications to large-scale integrated circuits.

Besides applications to flexible electronics, MoS₂ can be applied to flexible optoelectronics, such as photodetectors, by

utilizing its high photoabsorption coefficient.^{52,53} A 10×10 array of ultrathin phototransistors is fabricated (Figure S17a). The transfer curves ($I_{ds}-V_g$) of a representative phototransistor illuminated by 532 nm light is shown in Figure 5l. The time-resolved photocurrent responses are measured at $V_g = -10 \text{ V}$ by gradually increasing light intensities, showing light-sensitive current responses (Figure 5m). Typically, the MoS₂ phototransistor features residual photoconductivity due to charge trapping, even after the light illumination is removed.⁵⁴ Therefore, trapped charges are released every 150 s by applying a positive gate bias to maintain a low level of dark current ($\sim 1 \text{ nA}$). The ratio of photocurrent (I_{ph}) to dark current (I_{dark}) is proportional to the intensity of the illuminated light (Figure 5n). The photoresponsivity is 556 A/W at a light intensity of 0.11 mW/cm^2 (Figure S17b). I_{dark} and I_{ph} of 100 devices in the array are measured under a light intensity of 0.64 mW/cm^2 (Figure 5o), whose mean values are 1.52 nA and 1.31 μA , respectively, and thus I_{ph}/I_{dark} is $\sim 10^3$.

We develop a wafer-scale controlled synthetic process of TMD monolayers, which utilizes transition metal oxide nanoparticles as precursors. Facile deposition of precursors with uniform coverage on the inch-scale substrate facilitates the reproducible and massive production of TMD monolayers. The use of different mixing ratios of the dual-precursor, such as MoO₂ and W₁₈O₄₉ nanoparticles, enables one to manipulate the composition of the Mo_{1-x}W_xS₂ alloys, which allows the wafer-scale production of 2D materials with a tunable optical bandgap. Furthermore, our method can be widely applied to the mass production of various 2D materials, with controllability in terms of the coverage, morphology, and doping level. Ultrathin field-effect transistors and flexible electronic devices with uniform performance are fabricated by utilizing the continuous MoS₂ film, which demonstrates versatile device applications of TMD monolayers obtained from the new synthetic process.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.nanolett.1c02991>.

Experimental section and supporting data such as TMD growth, characterization, DFT calculation, and device fabrication (PDF)

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J.K. (Jihoon Kim) and J.P. planned the research. J.K. (Jihoon Kim), H.S., S.K., B.K.C., J.S.K., T.H., S.S., and J.P. discussed and/or acquired the data. H.S., C.C., and D.-H.K. fabricated the ultrathin flexible electronics and transistors. D.K., H.B., and H.L. performed the DFT calculation. J.K. (Jihoon Kim), J.K. (Joodeok Kim), and H.P. conducted the image processing. J.K. (Jihoon Kim), H.S., D.-H.K., and J.P. wrote the manuscript. D.-H.K., S.S., and J.P. supervised the research. All authors contributed to the discussion of results.

Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) Radisavljevic, B.; Radenovic, A.; Brivio, J.; Giacometti, V.; Kis, A. Single-layer MoS₂ Transistors. *Nat. Nanotechnol.* **2011**, *6*, 147–150.
- (2) Shi, W.; Kahn, S.; Jiang, L.; Wang, S.-Y.; Tsai, H.-Z.; Wong, D.; Taniguchi, T.; Watanabe, K.; Wang, F.; Crommie, M. F.; Zettl, A. Reversible Writing of High-Mobility and High-Carrier-Density Doping Patterns in Two-Dimensional van der Waals Heterostructures. *Nat. Electron.* **2020**, *3*, 99–105.
- (3) Karunadasa, H. I.; Montalvo, E.; Sun, Y.; Majda, M.; Long, J. R.; Chang, C. J. A Molecular MoS₂ Edge Site Mimic for Catalytic Hydrogen Generation. *Science* **2012**, *335*, 698–702.
- (4) Jin, C.; Regan, E. C.; Wang, D.; Iqbal Bakti Utama, M.; Yang, C.-S.; Cain, J.; Qin, Y.; Shen, Y.; Zheng, Z.; Watanabe, K.; Taniguchi, T.; Tongay, S.; Zettl, A.; Wang, F. Identification of Spin, Valley and Moiré Quasi-Angular Momentum of Interlayer Excitons. *Nat. Phys.* **2019**, *15*, 1140–1144.
- (5) Nayak, P. K.; Horbatenko, Y.; Ahn, S.; Kim, G.; Lee, J.-U.; Ma, K. Y.; Jang, A.-R.; Lim, H.; Kim, D.; Ryu, S.; Cheong, H.; Park, N.; Shin, H. S. Probing Evolution of Twist-Angle-Dependent Interlayer Excitons in MoSe₂/WSe₂ van der Waals Heterostructures. *ACS Nano* **2017**, *11*, 4041–4050.
- (6) Mak, K. F.; McGill, K. L.; Park, J.; McEuen, P. L. The Valley Hall Effect in MoS₂ Transistors. *Science* **2014**, *344*, 1489–1492.
- (7) Mak, K. F.; Lee, C.; Hone, J.; Shan, J.; Heinz, T. F. Atomically Thin MoS₂: A New Direct-Gap Semiconductor. *Phys. Rev. Lett.* **2010**, *105*, 136805.
- (8) Yin, X.; Ye, Z.; Chenet, D. A.; Ye, Y.; O'Brien, K.; Hone, J. C.; Zhang, X. Edge Nonlinear Optics on a MoS₂ Atomic Monolayer. *Science* **2014**, *344*, 488–490.

- (9) Voiry, D.; Fullon, R.; Yang, J.; de Carvalho Castro e Silva, C.; Kappera, R.; Bozkurt, I.; Kaplan, D.; Lagos, M. J.; Batson, P. E.; Gupta, G.; Mohite, A. D.; Dong, L.; Er, D.; Shenoy, V. B.; Asefa, T.; Chhowalla, M. The Role of Electronic Coupling between Substrate and 2D MoS₂ Nanosheets in Electrocatalytic Production of Hydrogen. *Nat. Mater.* **2016**, *15*, 1003–1009.
- (10) Zhang, J.; Yu, H.; Chen, W.; Tian, X.; Liu, D.; Cheng, M.; Xie, G.; Yang, W.; Yang, R.; Bai, X.; Shi, D.; Zhang, G. Scalable Growth of High-Quality Polycrystalline MoS₂ Monolayers on SiO₂ with Tunable Grain Sizes. *ACS Nano* **2014**, *8*, 6024–6030.
- (11) Shim, J.; Bae, S.-H.; Kong, W.; Lee, D.; Qiao, K.; Nezich, D.; Park, Y. J.; Zhao, R.; Sundaram, S.; Li, X.; Yeon, H.; Choi, C.; Kum, H.; Yue, R.; Zhou, G.; Ou, Y.; Lee, K.; Moodera, J.; Zhao, X.; Ahn, J.-H.; Hinkle, C.; Ougazzaden, A.; Kim, J. Controlled Crack Propagation for Atomic Precision Handling of Wafer-Scale Two-Dimensional Materials. *Science* **2018**, *362*, 665–670.
- (12) Yin, Z.; Li, H.; Li, H.; Jiang, L.; Shi, Y.; Sun, Y.; Lu, G.; Zhang, Q.; Chen, X.; Zhang, H. Single-Layer MoS₂ Phototransistors. *ACS Nano* **2012**, *6*, 74–80.
- (13) Coleman, J. N.; Lotya, M.; O'Neill, A.; Bergin, S. D.; King, P. J.; Khan, U.; Young, K.; Gaucher, A.; De, S.; Smith, R. J.; Shvets, I. V.; Arora, S. K.; Stanton, G.; Kim, H.-Y.; Lee, K.; Kim, G. T.; Duesberg, G. S.; Hallam, T.; Boland, J. J.; Wang, J. J.; Donegan, J. F.; Grunlan, J. C.; Moriarty, G.; Shmeliov, A.; Nicholls, R. J.; Perkins, J. M.; Grievson, E. M.; Theuwissen, K.; McComb, D. W.; Nellist, P. D.; Nicolosi, V. Two-Dimensional Nanosheets Produced by Liquid Exfoliation of Layered Materials. *Science* **2011**, *331*, 568–571.
- (14) van der Zande, A. M.; Huang, P. Y.; Chenet, D. A.; Berkelbach, T. C.; You, Y.; Lee, G.-H.; Heinz, T. F.; Reichman, D. R.; Muller, D. A.; Hone, J. C. Grains and Grain Boundaries in Highly Crystalline Monolayer Molybdenum Disulfide. *Nat. Mater.* **2013**, *12*, 554–561.
- (15) Kang, K.; Xie, S.; Huang, L.; Han, Y.; Huang, P. Y.; Mak, K. F.; Kim, C.-J.; Muller, D.; Park, J. High-Mobility Three-Atom-Thick Semiconducting Films with Wafer-Scale Homogeneity. *Nature* **2015**, *520*, 656–660.
- (16) Cohen, A.; Patsha, A.; Mohapatra, P. K.; Kazes, M.; Ranganathan, K.; Houben, L.; Oron, D.; Ismach, A. Growth-Etch Metal-Organic Chemical Vapor Deposition Approach of WS₂ Atomic Layers. *ACS Nano* **2021**, *15*, 526–538.
- (17) Wang, J.; Cai, X.; Shi, R.; Wu, Z.; Wang, W.; Long, G.; Tang, Y.; Cai, N.; Ouyang, W.; Geng, P.; Chandrashekar, B. N.; Amini, A.; Wang, N.; Cheng, C. Twin Defect Derived Growth of Atomically Thin MoS₂ Dendrites. *ACS Nano* **2018**, *12*, 635–643.
- (18) Xu, Z.-Q.; Zhang, Y.; Lin, S.; Zheng, C.; Zhong, Y. L.; Xia, X.; Li, Z.; Sophia, P. J.; Fuhrer, M. S.; Cheng, Y.-B.; Bao, Q. Synthesis and Transfer of Large-Area Monolayer WS₂ Crystals: Moving Toward the Recyclable Use of Sapphire Substrates. *ACS Nano* **2015**, *9*, 6178–6187.
- (19) Dou, J.; Zeng, H. C. Targeted Synthesis of Silicomolybdenic Acid (Keggin Acid) inside Mesoporous Silica Hollow Spheres for Friedel-Crafts Alkylation. *J. Am. Chem. Soc.* **2012**, *134*, 16235–16246.
- (20) Qiu, J.; Xiao, Q.; Zheng, X.; Zhang, L.; Xing, H.; Ni, D.; Liu, Y.; Zhang, S.; Ren, Q.; Hua, Y.; Zhao, K.; Bu, W. Single W₁₈O₄₉ Nanowires: A Multifunctional Nanoplatfor for Computed Tomography Imaging and Photothermal/Photodynamic/Radiation Synergistic Cancer Therapy. *Nano Res.* **2015**, *8*, 3580–3590.
- (21) Li, H.; Zhang, Q.; Yap, C. C. R.; Tay, B. K.; Edwin, T. H. T.; Olivier, A.; Baillargeat, D. From Bulk to Monolayer MoS₂: Evolution of Raman Scattering. *Adv. Funct. Mater.* **2012**, *22*, 1385–1390.
- (22) Splendiani, A.; Sun, L.; Zhang, Y.; Li, T.; Kim, J.; Chim, C.-Y.; Galli, G.; Wang, F. Emerging Photoluminescence in Monolayer MoS₂. *Nano Lett.* **2010**, *10*, 1271–1275.
- (23) Ji, Q.; Zhang, Y.; Gao, T.; Zhang, Y.; Ma, D.; Liu, M.; Chen, Y.; Qiao, X.; Tan, P.-H.; Kan, M.; Feng, J.; Sun, Q.; Liu, Z. Epitaxial Monolayer MoS₂ on Mica with Novel Photoluminescence. *Nano Lett.* **2013**, *13*, 3870–3877.
- (24) Suenaga, K.; Ji, H. G.; Lin, Y.-C.; Vincent, T.; Maruyama, M.; Aji, A. S.; Shiratsuchi, Y.; Ding, D.; Kawahara, K.; Okada, S.; Panchal, V.; Kazakova, O.; Hibino, H.; Suenaga, K.; Ago, H. Surface-Mediated Aligned Growth of Monolayer MoS₂ and In-Plane Heterostructures with Graphene on Sapphire. *ACS Nano* **2018**, *12* (10), 10032–10044.
- (25) Li, S.; Lin, Y.-C.; Zhao, W.; Wu, J.; Wang, Z.; Hu, Z.; Shen, Y.; Tang, D.-M.; Wang, J.; Zhang, Q.; Zhu, H.; Chu, L.; Zhao, W.; Liu, C.; Sun, Z.; Taniguchi, T.; Osada, M.; Chen, W.; Xu, Q.-H.; Wee, A. T. S.; Suenaga, K.; Ding, F.; Eda, G. Vapour-Liquid-Solid Growth of Monolayer MoS₂ Nanoribbons. *Nat. Mater.* **2018**, *17*, 535–542.
- (26) Li, J.; Yan, W.; Lv, Y.; Zhao, J.; Zhang, D.; O'Coilain, C.; Cullen, C. P.; Stimpel-Lindner, T.; Duesberg, G. S.; Cho, J.; Choi, M.; Chun, B. S.; Zhao, Y.; Lv, C.; Arora, S. K.; Wu, H.-C. Sub-Millimeter Size High Mobility Single Crystal MoSe₂ Monolayers Synthesized by NaCl-assisted Chemical Vapor Deposition. *RSC Adv.* **2020**, *10*, 1580–1587.
- (27) Yang, P.; Zou, X.; Zhang, Z.; Hong, M.; Shi, J.; Chen, S.; Shu, J.; Zhao, L.; Jiang, S.; Zhou, X.; Huan, Y.; Xie, C.; Gao, P.; Chen, Q.; Zhang, Q.; Liu, Z.; Zhang, Y. Batch Production of 6-Inch Uniform Monolayer Molybdenum Disulfide Catalyzed by Sodium in Glass. *Nat. Commun.* **2018**, *9*, 979.
- (28) Zhou, W.; Zou, X.; Najmaei, S.; Liu, Z.; Shi, Y.; Kong, J.; Lou, J.; Ajayan, P. M.; Yakobson, B. I.; Idrobo, J.-C. Intrinsic Structural Defects in Monolayer Molybdenum Disulfide. *Nano Lett.* **2013**, *13* (6), 2615–2622.
- (29) Zhou, S.; Wang, S.; Shi, Z.; Sawada, H.; Kirkland, A. I.; Li, J.; Warner, J. H. Atomically Sharp Interlayer Stacking Shifts at Antiphase Grain Boundaries in Overlapping MoS₂ Secondary Layers. *Nanoscale* **2018**, *10*, 16692–16702.
- (30) Ly, T. H.; Perello, D. J.; Zhao, J.; Deng, Q.; Kim, H.; Han, G. H.; Chae, S. H.; Jeong, H. Y.; Lee, Y. H. Misorientation-Angle-Dependent Electrical Transport across Molybdenum Disulfide Grain Boundaries. *Nat. Commun.* **2016**, *7*, 10426.
- (31) Zhou, J.; Lin, J.; Huang, X.; Zhou, Y.; Chen, Y.; Xia, J.; Wang, H.; Xie, Y.; Yu, H.; Lei, J.; Wu, D.; Liu, F.; Fu, Q.; Zeng, Q.; Hsu, C.-H.; Yang, C.; Lu, L.; Yu, T.; Shen, Z.; Lin, H.; Yakobson, B. I.; Liu, Q.; Suenaga, K.; Liu, G.; Liu, Z. A Library of Atomically Thin Metal Chalcogenides. *Nature* **2018**, *556*, 355–359.
- (32) Susarla, S.; Kutana, A.; Hachtel, J. A.; Kochat, V.; Apte, A.; Vajtai, R.; Idrobo, J. C.; Yakobson, B. I.; Tiwary, C. S.; Ajayan, P. M. Quaternary 2D Transition Metal Dichalcogenides (TMDs) with Tunable Bandgap. *Adv. Mater.* **2017**, *29*, 1702457.
- (33) Duan, X.; Wang, C.; Fan, Z.; Hao, G.; Kou, L.; Halim, U.; Li, H.; Wu, X.; Wang, Y.; Jiang, J.; Pan, A.; Huang, Y.; Yu, R.; Duan, X. Synthesis of WS₂xSe_{2-2x} Alloy Nanosheets with Composition-Tunable Electronic Properties. *Nano Lett.* **2016**, *16*, 264–269.
- (34) Zhang, T.; Fujisawa, K.; Zhang, F.; Liu, M.; Lucking, M. C.; Gontijo, R. N.; Lei, Y.; Liu, H.; Crust, K.; Granzier-Nakajima, T.; Terrones, H.; Elias, A. L.; Terrones, M. Universal *In Situ* Substitutional Doping of Transition Metal Dichalcogenides by Liquid-Phase Precursor-Assisted Synthesis. *ACS Nano* **2020**, *14*, 4326–4335.
- (35) Duan, X.; Wang, C.; Shaw, J. C.; Cheng, R.; Chen, Y.; Li, H.; Wu, X.; Tang, Y.; Zhang, Q.; Pan, A.; Jiang, J.; Yu, R.; Huang, Y.; Duan, X. Lateral Epitaxial Growth of Two-Dimensional Layered Semiconductor Heterojunctions. *Nat. Nanotechnol.* **2014**, *9*, 1024–1030.
- (36) Yang, J.; Mohamad, A. R.; Wang, Y.; Fullon, R.; Song, X.; Zhao, F.; Bozkurt, I.; Augustin, M.; Santos, E. J. G.; Shin, H. S.; Zhang, W.; Voiry, D.; Jeong, H. Y.; Chhowalla, M. Ultrahigh-Current-Density Niobium Disulfide Catalysts for Hydrogen Evolution. *Nat. Mater.* **2019**, *18*, 1309–1314.
- (37) Zhou, J.; Tang, B.; Lin, J.; Lv, D.; Shi, J.; Sun, L.; Zeng, Q.; Niu, L.; Liu, F.; Wang, X.; Liu, X.; Suenaga, K.; Jin, C.; Liu, Z. Morphology Engineering in Monolayer MoS₂-WS₂ Lateral Heterostructures. *Adv. Funct. Mater.* **2018**, *28*, 1801568.
- (38) Liang, L.; Meunier, V. First-Principles Raman Spectra of MoS₂, WS₂ and Their Heterostructures. *Nanoscale* **2014**, *6*, 5394–5401.
- (39) Hong, X.; Kim, J.; Shi, S.-F.; Zhang, Y.; Jin, C.; Sun, Y.; Tongay, S.; Wu, J.; Zhang, Y.; Wang, F. Ultrafast Charge Transfer in Atomically Thin MoS₂/WS₂ Heterostructures. *Nat. Nanotechnol.* **2014**, *9*, 682–686.

(40) Zheng, S.; Sun, L.; Yin, T.; Dubrovkin, A. M.; Liu, F.; Liu, Z.; Shen, Z. X.; Fan, H. J. Monolayers of $W_xMo_{1-x}S_2$ Alloy Heterostructure with In-Plane Composition Variations. *Appl. Phys. Lett.* **2015**, *106*, 063113.

(41) Chen, Y.; Xi, J.; Dumcenco, D. O.; Liu, Z.; Suenaga, K.; Wang, D.; Shuai, Z.; Huang, Y.-S.; Xie, L. Tunable Band Gap Photoluminescence from Atomically Thin Transition-Metal Dichalcogenide Alloys. *ACS Nano* **2013**, *7*, 4610–4616.

(42) Gurarlan, A.; Yu, Y.; Su, L.; Yu, Y.; Suarez, F.; Yao, S.; Zhu, Y.; Ozturk, M.; Zhang, Y.; Cao, L. Surface-Energy-Assisted Perfect Transfer of Centimeter-Scale Monolayer and Few-Layer MoS_2 Films onto Arbitrary Substrates. *ACS Nano* **2014**, *8*, 11522–11528.

(43) Cain, J. D.; Shi, F.; Wu, J.; Dravid, V. P. Growth Mechanism of Transition Metal Dichalcogenide Monolayers: The Role of Self-Seeding Fullerene Nuclei. *ACS Nano* **2016**, *10*, 5440–5445.

(44) Li, Y.; Hao, S.; DiStefano, J. G.; Murthy, A. A.; Hanson, E. D.; Xu, Y.; Wolverson, C.; Chen, X.; Dravid, V. P. Site-Specific Positioning and Patterning of MoS_2 Monolayers: The Role of Au Seeding. *ACS Nano* **2018**, *12*, 8970–8976.

(45) Shi, J.; Ma, D.; Han, G.-F.; Zhang, Y.; Ji, Q.; Gao, T.; Sun, J.; Song, X.; Li, C.; Zhang, Y.; Lang, X.-Y.; Zhang, Y.; Liu, Z. Controllable Growth and Transfer of Monolayer MoS_2 on Au Foils and Its Potential Application in Hydrogen Evolution Reaction. *ACS Nano* **2014**, *8*, 10196–10204.

(46) An, Y.-R.; Fan, X.-L.; Luo, Z.-F.; Lau, W.-M. Nanopolygons of Monolayer MS_2 : Best Morphology and Size for HER Catalysis. *Nano Lett.* **2017**, *17*, 368–376.

(47) Mun, J.; Park, H.; Park, J.; Joung, D.; Lee, S.-K.; Leem, J.; Myoung, J.-M.; Park, J.; Jeong, S.-H.; Chegal, W.; Nam, S.; Kang, S.-W. High-Mobility MoS_2 Directly Grown on Polymer Substrate with Kinetics-Controlled Metal-Organic Chemical Vapor Deposition. *ACS Appl. Electron. Mater.* **2019**, *1*, 608–616.

(48) Lu, J.; Yeo, P. S. E.; Gan, C. K.; Wu, P.; Loh, K. P. Transforming C_{60} Molecules into Graphene Quantum Dots. *Nat. Nanotechnol.* **2011**, *6*, 247–252.

(49) Coraux, J.; N'Diaye, A. T.; Engler, M.; Busse, C.; Wall, D.; Buckanie, N.; Meyer zu Heringdorf, F.-J.; van Gestel, R.; Poelsema, B.; Michely, T. Growth of Graphene on Ir(111). *New J. Phys.* **2009**, *11*, 023006.

(50) Wang, Q.; Li, N.; Tang, J.; Zhu, J.; Zhang, Q.; Jia, Q.; Lu, Y.; Wei, Z.; Yu, H.; Zhao, Y.; Guo, Y.; Gu, L.; Sun, G.; Yang, W.; Yang, R.; Shi, D.; Zhang, G. Wafer-Scale Highly Oriented Monolayer MoS_2 with Large Domain Sizes. *Nano Lett.* **2020**, *20*, 7193–7199.

(51) Li, N.; Wang, Q.; Shen, C.; Wei, Z.; Yu, H.; Zhao, J.; Lu, X.; Wang, G.; He, C.; Xie, L.; Zhu, J.; Du, L.; Yang, R.; Shi, D.; Zhang, G. Large-Scale Flexible and Transparent Electronics Based on Monolayer Molybdenum Disulfide Field-Effect Transistors. *Nat. Electron.* **2020**, *3*, 711–717.

(52) Choi, C.; Choi, M. K.; Liu, S.; Kim, M. S.; Park, O. K.; Im, C.; Kim, J.; Qin, X.; Lee, G. J.; Cho, K. W.; Kim, M.; Joh, E.; Lee, J.; Son, D.; Kwon, S.-H.; Jeon, N. L.; Song, Y. M.; Lu, N.; Kim, D.-H. Human Eye-Inspired Soft Optoelectronic Device Using High-Density MoS_2 -Graphene Curved Image Sensor Array. *Nat. Commun.* **2017**, *8*, 1664.

(53) Choi, C.; Leem, J.; Kim, M. S.; Taqieddin, A.; Cho, C.; Cho, K. W.; Lee, G. J.; Seung, H.; Bae, H. J.; Song, Y. M.; Hyeon, T.; Aluru, N. R.; Nam, S.; Kim, D.-H. Curved Neuromorphic Image Sensor Array Using a MoS_2 -Organic Heterostructure Inspired by The Human Visual Recognition System. *Nat. Commun.* **2020**, *11*, 5934.

(54) Di Bartolomeo, A.; Genovese, L.; Foller, T.; Giubileo, F.; Luongo, G.; Croin, L.; Liang, S.-J.; Ang, L. K.; Schleberger, M. Electrical Transport and Persistent Photoconductivity in Monolayer MoS_2 Phototransistors. *Nanotechnology* **2017**, *28*, 214002.